## Alkaloids from Fruits of Daphniphyllum oldhamii

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Seventeen Daphniphyllum alkaloids, including two new pentacyclic alkaloids, yuzuric acid (1) and daphnezomic acid (2), and 15 known ones, were isolated from the fruits of Daphniphyllum oldhamii. The structures and configuration of the two new alkaloids were determined on the basis of spectroscopic methods, especially 2D NMR techniques.

Introduction. – The genus *Daphniphyllum* (Daphniphyllaceae) comprises about 30 plant species distributed mainly in the southeast of Asia. Ten of them were found in southern China [1]. The highly complex Daphniphyllum alkaloids have been the attracting research programs of natural-products chemistry. A series of structurally diverse Daphniphyllum alkaloids have been isolated from different plant parts of Daphniphyllum species in our research group [2].

Daphniphyllum oldhamii (HEMSL.) ROSENTH. is a small evergreen tree. Its leaves and roots were applied in traditional Chinese medicine for the treatment of fever, snakebite, and fractures [3]. Our previous chemical investigation [4] on the leaves of D. oldhamii showed the presence of six polycyclic alkaloids, and subsequent studies on its aerial parts of fresh saplings and roots by Hao and co-workers [5] resulted in the isolation of 13 Daphniphyllum alkaloids. The fruits of this plant have not been chemically studied previously. In the present study, two new pentacyclic Daphniphyl*lum* alkaloids, yuzuric acid (1) and daphnezomic acid (2), together with 15 known ones, daphnigraciline, daphnezomine R, daphnigracine, a zwitterionic alkaloid ( $O^{20}$ -deacetyl-5-de(acetyloxy)-8,9-didehydro-2-deoxy-8,12-dihydro-9,22-secoyuzurimin-23-oic acid), daphnezomine S, yuzurimic acid B, daphnilactone B and its methyl ester, daphnezomine H, yuzurimine B, paxdaphnines A and B, deoxyisocalyciphylline B, deoxycalyciphylline B, and calyciphylline B were isolated from the EtOH extract of the fruits of D. oldhamii. The structures of the new alkaloids were elucidated on the basis of spectroscopic methods, especially 2D NMR techniques.

**Results and Discussion.** – Alkaloid **1** was isolated as a white amorphous solid. Its molecular formula, C23H35NO4, was determined on the basis of the molecular-ion peak at m/z 389.2545 in the HR-EI-MS. The IR spectrum showed absorption bands between 3000 and 2500 cm<sup>-1</sup>, and at 1711 cm<sup>-1</sup>, typical for the presence of a carboxylic acid. The <sup>1</sup>H- and <sup>13</sup>C-NMR (*Table*), HMBC (*Fig. 1*), and ROESY data (*Fig. 2*) showed that

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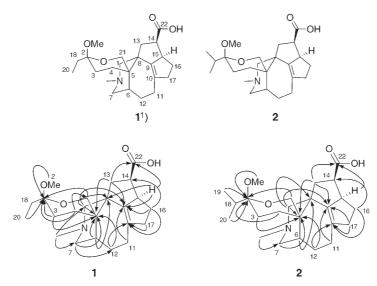


Fig. 1. Key HMBC correlations  $(\mathrm{H}\,{\rightarrow}\,\mathrm{C})$  of compounds 1 and 2

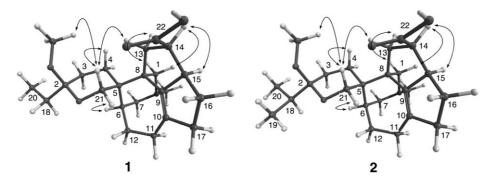


Fig. 2. 3D Structures and key ROESY correlations  $({\rm H} \,{\leftrightarrow}\, {\rm H})$  of compounds 1 and 2

compound **1** was the ester hydrolysate of yuzurine [6]. Compound **1** was named yuzuric acid.

The <sup>1</sup>H-NMR spectrum of **1** showed two geminal protons at  $\delta(H) 4.25$  (d, J = 12.8 Hz) and 3.82 (dd, J = 12.8, 2.2 Hz) assignable to CH<sub>2</sub>(21), which is diagnostic for a yuzurine-type skeleton of a *Daphniphyllum* alkaloid. The resonances at  $\delta(H) 3.19$  (s) and 2.73 (s) were attributed to an MeO and a MeN group, respectively. A signal at  $\delta(H) 0.85$  (t, J = 7.4 Hz) was assigned to Me(20), indicating that an Et group was attached to C(2). The <sup>13</sup>C-NMR (with DEPT) spectrum confirmed the yuzurine-type skeleton of **1**. The signals at  $\delta(C) 146.1$  (s, C(9)) and 136.6 (s, C(10)) stem from a tetrasubstituted C=C bond. The resonance at  $\delta(C) 100.8$  (s, C(2)) revealed the presence of the ketal group. A C=O signal at  $\delta(C) 181.2$  (s, C(22)) confirmed the presence of a COOH group. Except for the presence of a C=C bond and the COOH group, 7 degrees of unsaturation in the molecule required a pentacyclic skeleton for

<sup>1)</sup> Trivial numbering; for systematic names, see Exper. Part.

	1		2	
	$\delta$ (H)	δ (C)	δ(H)	δ (C)
CH <sub>2</sub> (1)	2.95 (br. s)	59.7 (t)	2.73 $(d, J = 12.2, H_a),$ 2.67 $(d, J = 12.2, H_{\beta})$	60.7 ( <i>t</i> )
C(2)		100.8 (s)	$2107 (a, 0 1202, 11_p)$	102.7(s)
CH <sub>2</sub> (3)	1.64–1.67 ( <i>m</i> )	29.3 <i>(t)</i>	$1.72 - 1.76 (m, H_a),$ $1.36 (br. d, 10.3, H_b)$	23.4 <i>(t)</i>
CH <sub>2</sub> (4)	$1.92 - 1.98 (m, H_{a}),$ $1.69 - 1.71 (m, H_{\beta})$	24.1 <i>(t)</i>	1.94 – 1.96 $(m, H_{\alpha})$ , 1.66 – 1.69 $(m, H_{\beta})$	23.7 <i>(t)</i>
C(5)		37.9 (s)		37.9 (s)
H-C(6)	2.55 (br. s)	34.4(d)	2.33 - 2.37 (m)	34.5 (d)
CH <sub>2</sub> (7)	$3.30-3.32 (m, H_a),$ $3.24 (dd, J = 13.1, 5.4, 1 H_b)$	56.7 ( <i>t</i> )	3.09 (br. $d, J = 12.8, H_a$ ), 2.97–2.99 ( $m, H_b$ )	57.1 <i>(t)</i>
C(8)		47.9(s)		48.1 (s)
C(9)		146.1 (s)		147.2 (s)
C(10)		136.6 (s)		135.4 (s)
CH <sub>2</sub> (11)	$2.35 - 2.42 (m, H_a),$ $2.24 - 2.29 (m, H_b)$	27.5 <i>(t)</i>	$2.39 - 2.44 (m, H_a),$ 2.20 (dd, J = 16.2, 5.4, H <sub>b</sub> )	27.9 ( <i>t</i> )
$CH_{2}(12)$	1.74 - 1.76(m)	27.5(t)	1.68 - 1.71 (m)	28.0(t)
CH <sub>2</sub> (13)	1.61 – 1.64 $(m, H_{\alpha})$ , 2.82 – 2.87 $(m, H_{\beta})$	41.4 <i>(t)</i>	$1.55 - 1.58 (m, H_a),$ $2.80 - 2.85 (m, H_{\beta})$	41.7 <i>(t)</i>
H - C(14)	3.48 (br. s)	56.3 (d)	3.44 (br. s)	56.4(d)
H - C(15)	2.76 - 2.79(m)	46.0(d)	2.74 - 2.77(m)	46.6 (d)
CH <sub>2</sub> (16)	$1.88 - 1.93 (m, H_a),$ $1.53 - 1.61 (m, H_b)$	30.1 <i>(t)</i>	$1.86 - 1.90 (m, H_a),$ $1.57 - 1.62 (m, H_b)$	30.4 <i>(t)</i>
CH <sub>2</sub> (17)	$2.60-2.67 (m, H_a),$ $2.32-2.38 (m, H_b)$	44.1 <i>(t)</i>	$2.55-2.60 (m, H_a),$ $2.29-2.35 (m, H_b)$	44.0 <i>(t)</i>
CH <sub>2</sub> (18) or H–C(18)	$1.71 - 1.74 (m, H_a),$ $1.39 - 1.46 (m, H_b)$	30.1 ( <i>t</i> )	2.01–2.08 ( <i>m</i> , 1 H)	33.0 ( <i>d</i> )
Me(19)	_	_	0.92 (d, J = 6.4)	18.2(q)
Me(20)	0.85 (t, 7.4)	8.5(q)	0.84 (d, J = 7.0)	17.3(q)
CH <sub>2</sub> (21)	3.82 ( $dd$ , $J = 12.8, 2.2, H_a$ ), 4.25 ( $d$ , $J = 12.8, H_\beta$ )	64.3(t)	3.86 $(dd, J = 12.5, 2.9, H_a),$ 4.21 $(d, J = 12.9, H_{\beta})$	64.7 ( <i>t</i> )
C(22)		181.2(s)	· · · · · · · · · · · · · · · · · · ·	182.3 (s)
Me-O	3.19(s)	48.3(q)	3.18(s)	47.5 (q)
MeN	2.73(s)	46.6(q)	2.52(s)	47.0(q)

Table. <sup>1</sup>*H*- and <sup>13</sup>*C*-*NMR* (with DEPT) Data (CD<sub>3</sub>OD, 400 and 100 MHz), resp. of Alkaloids 1 and  $2^1$ ).  $\delta$  in ppm, J in Hz.

alkaloid **1**. A HMBC experiment confirmed the constitution of **1** (*Fig. 1*), in which the key correlations  $CH_2(13)/C(22)$  and H-C(15)/C(22) revealed that COOH was located at C(14); the correlations of MeO/C(2), Me(20)/C(2), and  $CH_2(21)/C(2)$  confirmed the presence of the ketal group and the location of the Et group; the position of the MeN group was confirmed by the correlations MeN/C(1) and MeN/C(7). The C(9)=C(10) bond was established by the mutual HMBC correlations  $CH_2(13)/C(9)$ ,  $CH_2(12)/C(10)$ ,  $CH_2(16)/C(10)$ , and  $CH_2(17)/C(10)$ . The HMBC cross-peaks  $CH_2(13)/C(8)$ ,  $CH_2(11)/C(9)$ ,  $CH_2(7)/C(12)$  further supported the proposed constitutional formula of **1**. The relative configuration of **1** was determined to be the same as in yuzurine by the ROESY correlations  $CH_2(1)/H-C(14), H-C(14)/H-C(15)$ , and  $CH_2(1)/H_a-C(13)$  which showed that these protons were on the same side of the ring and were arbitrarily assigned to be  $\alpha$ -oriented. The ROESY correlations of

MeO/H<sub> $\beta$ </sub>-C(21), H<sub> $\beta$ </sub>-C(21)/H<sub> $\beta$ </sub>-C(4) showed that they were on the other side of the yuzurine-type ring system and were thus assigned the  $\beta$ -configuration.

Alkaloid **2** was obtained as a white amorphous solid. The HR-EI-MS molecular-ion peak at m/z 403.2725 revealed its molecular formula as  $C_{24}H_{37}NO_4$ , which is 14 mass units more than **1**, and 14 mass units less than daphnezomine R [7]. The IR spectrum also exhibited typical absorption bands between 3000 and 2500 cm<sup>-1</sup> and at 1705 cm<sup>-1</sup> indicating the presence of a carboxylic acid. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of **2** (*Table*) showed high similarity to those of compound **1**. Comparison with the <sup>1</sup>H-NMR data of daphnezomine R indicated that **2** was likely the ester hydrolysate of daphnezomine R [7]. The HMBC and ROESY data (*Figs. 1* and 2) confirmed the proposed structure of **2** which was named daphnezomic acid.

In the <sup>1</sup>H-NMR spectrum of **2**, two Me groups resonated at  $\delta(H) 0.92$  (d, J = 6.4 Hz) and 0.84 (d, J = 7.0 Hz) and a CH group at  $\delta(H) 2.06$  (m, 1 H), typical of an isopropyl group, which was tentatively placed at C(2). Compared to daphnezomine R, the MeO, signal of the COOMe group was absent in the <sup>1</sup>H-NMR spectrum of **2**. The HMBC plot revealed the key correlations Me(19)/C(2), Me(20)/C(2), MeO/C(2), CH<sub>2</sub>(21)/C(2), and CH<sub>2</sub>(3)/C(2), showing the existence of and <sup>i</sup>Pr, MeO, and ketal group and their connectivity (*Fig. 1*). The presence of a COOH group at C(14) was confirmed by the correlations CH<sub>2</sub>(13)/C(22) and H–C(15)/C(22), and the correlations CH<sub>2</sub>(1)/C(9) and CH<sub>2</sub>(12)/C(10) further indicated the presence of a C(9)=C(10) bond. Thus, the constitution of **2** was elucidated. The relative configuration of **2** was identical with that of daphnezomine R as established by a ROESY experiment, in which the key correlations H<sub>a</sub>–C(1)/H–C(14), H–C(15), H<sub>b</sub>–C(1)/H<sub>a</sub>–C(13), H–C(6)/H<sub>a</sub>–C(21), H<sub>b</sub>–C(13)/H<sub>b</sub>–C(21), and MeO/H<sub>b</sub>–C(21) were observed (*Fig. 2*).

The chemical shifts in the vicinity of the N-atom of alkaloids **1** and **2** as compared with those of their corresponding methyl esters [6][7], and also their IR absorptions clearly indicated that yuzuric acid (**1**) and daphnezomic acid (**2**) were present as shown rather than in the form of the corresponding zwitterions.

On the basis of the <sup>1</sup>H- and <sup>13</sup>C-NMR and MS data, the fifteen remaining alkaloids were shown to be already known and identified as daphnigraciline [6][8], daphnezomine R [7], daphnigracine [6][8], a zwitterionic alkaloid [9] ( $O^{20}$ -deacetyl-5de(acetyloxy)-8,9-didehydro-2-deoxy-8,12-dihydro-9,22-secoyuzurimin-23-oic acid), daphnezomine S[7], yuzurimic acid B [10], daphnilactone B and its methyl ester [11], daphnezomine H [11c], yuzurimine B [12], paxdaphnine A and B [13], deoxyisocalyciphylline B [2a], deoxycalyciphylline B [2a], and calyciphylline B [14].

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## **Experimental Part**

General. All solvents were of anal. grade (Shanghai Chemical Plant, Shanghai, People's Republic of China). Column chromatography (CC): silica gel (200–300 mesh),  $C_{18}$  reversed-phase silica gel (150–200 mesh; Merck), MCI gel (CHP20P, 75–150 µm; Mitsubishi Chemical Industries Ltd.), or Sephadex-LH-20 gel (Amersham Biosciences). TLC: precoated silica gel GF254 plates (Qingdao Haiyang Chemical Plant, Qingdao, People's Republic of China). Semi-prep. HPLC; Waters 515 pump; Waters 2487 detector; YMC-Pack-ODS-A column (250 × 10 mm, S-5 µm, 12 nm). Optical rotations:

*Perkin-Elmer 341* polarimeter. IR Spectra: *Perkin-Elmer 577* spectrometer; KBr disks, in cm<sup>-1</sup>. NMR Spectra: *Bruker AM-400, Varian Inova-400,* or *Varian Inova-600* spectrometer; SiMe<sub>4</sub> as internal standard. EI-MS (70 eV) and ESI-MS: *Finnigan MAT95* and *Finnigan LC-Q<sup>DECA</sup>* instrument, resp. in m/z (rel. %).

*Plant Material.* The fruits of *D. oldhamii* (HEMSL.) ROSENTH. was collected from Guangxi Province of P. R. China and authenticated by Prof. *Shao-Qing Tang*, Guangxi Normal University. A voucher specimen has been deposited at the Institute of Materia Medica, SIBS, Chinese Academy of Sciences (accession number: DO-T-frt-zg1Y).

Extraction and Isolation. The fresh fruits of D. oldhamii (4.9 kg) were extracted with 95% EtOH at r.t. to give a crude extract (135 g), which was dissolved in  $H_2O(11)$  to form a suspension and adjusted to pH ca. 3 with 2M  $H_2SO_4$ . The acidic suspension was first partitioned with petroleum ether (3 × 1 l) and AcOEt  $(3 \times 11)$  to remove the neutral components. The aq. phase was then basified to pH ca. 10 with sat.  $Na_2CO_3$  soln. and extracted with AcOEt (3 × 800 ml) to obtain 49 g of crude alkaloids. The latter were then subjected to CC (MCI): Fractions  $A_1 - A_3$ . Fr.  $A_1$  was further separated by CC (Sephadex-LH-20, 100% EtOH): Fr. A<sub>1a</sub> and A<sub>1b</sub>. Fr. A<sub>1a</sub> was then separated by CC (silica gel, AcOEt/EtOH/Et<sub>2</sub>NH 10:1:0.1) to give daphnilactone B (200 mg) and its methyl ester (15 mg), daphnezomine H (15 mg), as well as an alkaloid mixture, which was further purified by prep. HPLC (20% MeCN with 0.05% Et<sub>2</sub>NH): 1 (11 mg) and 2 (9 mg). Fr.  $A_{1b}$  was purified by prep. HPLC (8% MeCN with 0.05% Et<sub>2</sub>NH): yuzurimic acid B (8 mg). Fr. A2 was subjected to CC (Sephadex-LH-20, 100% EtOH): Fr. A2a and A2b. Fr. A2a was purified by prep. HPLC (12% MeCN): zwitter-ionic alkaloid (O20-deacetyl-5-de(acetyloxy)-8,9didehydro-2-deoxy-8,12-dihydro-9,22-secoyuzurimin-23-oic acid) (15 mg) and daphnezomine S (12 mg). Fr. A2b was separated by CC (silica gel, CH2Cl2/MeOH/Et2NH 30:1:0.1): paxdaphnine B (4 mg), yuzurimine B (9 mg), and daphnigracine (11 mg). Fr. A<sub>3</sub> was separated by CC (silica gel, AcOEt/ MeOH 25:1): Fr. A<sub>3a</sub>-A<sub>3d</sub>. Each of the latter was then purified by CC (amino silica gel, cyclohexane/ CHCl 20:1). Fr. A<sub>3a</sub> afforded daphnigraciline (7 mg) and daphnezomine R (9 mg), Fr. A<sub>3b</sub> deoxyisocalyciphylline B (9 mg) and deoxycalcyciphylline B (5 mg), Fr. A<sub>3c</sub> calyciphylline B (8 mg), and Fr. A<sub>3d</sub> paxdaphnine A (5 mg).

Yuzuric Acid (= rel-(3'R,4R,6'R,8aS,9S,10aS)-6'-Ethyl-2,3,4,5,5',6,6',7,8,8a,9,10-dodecahydro-6'-methoxy-2-methyl(spiro[1H-4,10a-methanopentaleno[1,6-cd]azonine-11,3'(4'H)-[2H]pyran]-9-carboxylic Acid; 1): White amorphous solid.  $[a]_D^{20} = -23.6$  (c = 0.59, MeOH). IR (KBr): 3431, 2937, 2833, 2779, 1711, 1633, 1570, 1466, 1379, 1180, 1043, 891. <sup>1</sup>H- and <sup>13</sup>C-NMR: *Table*. EI-MS (70 eV): 389 (4,  $M^+$ ), 374 (12), 357 (15), 329 (8), 300 (21), 273 (10), 58 (100). HR-EI-MS: 389. 2545 ( $M^+$ ,  $C_{23}H_{35}NO_4^+$ ; calc. 389.2566).

Daphnezomic Acid (=*rel*-(3'R,4R,6'S,8aS,9S,10aS)-2,3,4,5,5',6,6',7,8,8a,9,10-Dodecahydro-6'-methoxy-2-methyl-6'-(1-methylethyl)spiro[1H-4,10a-methanopentaleno[1,6-cd]azonine-11,3'(4H)-[2H]pyran]-9-carboxylic Acid; **2**): White amorphous solid.  $[a]_D^{20} = -24.9$  (c = 0.45, MeOH). IR (KBr): 3440, 2958, 2835, 1705, 1581, 1468, 1367, 1209, 1099, 1036, 899. <sup>1</sup>H- and <sup>13</sup>C-NMR: *Table*. EI-MS (70 eV): 403 (2,  $M^+$ ), 388 (16), 371 (30), 328 (12), 300 (36), 58(100). HR-EI-MS: 403.2725 ( $M^+$ ,  $C_{24}H_{37}NO_4$ ; calc. 403.2723).

## REFERENCES

- M. Zheng, T. L. Min, in 'Flora of China' ('Zhongguo Zhiwu Zhi'), Science Press, Beijing, 1980, Vol. (1), p. 1–11.
- [2] a) S. P. Yang, J. M. Yue, J. Org. Chem. 2003, 68, 7961; b) Z. J. Zhan, S. P. Yang, J. M. Yue, J. Org. Chem. 2004, 69, 1726; c) S. P. Yang, J. M. Yue, Org. Lett. 2004, 6, 1401; d) S. P. Yang, H. Zhang, C. R. Zhang, H. D. Chen, J. M. Yue, J. Nat. Prod. 2006, 69, 79; e) X. Chen, Z. J. Zhan, J. M. Yue, Helv. Chim. Acta 2005, 88, 854; f) H. Zhang, S. P. Yang, C. Q. Fan, J. Ding, J. M. Yue, J. Nat. Prod. 2006, 69, 553.
- [3] Editorial Committee of the Administration Bureau of Traditional Chinese Medicine, in 'Chinese Materia Medica' ('Zhonghua Bencao'), Shanghai Science & Technology Press, Shanghai, 1999, Vol. 4, p. 867.

- [4] X. Chen, Z. J. Zhan, J. M. Yue, Chem. Biodiv. 2004, 1, 1513.
- [5] a) S. Z. Mu, Y. Wang, H. P. He, X. W. Yang, Y. H. Wang, Y. T. Di, Y. Lu, Y. Chang, X. J. Hao, *J. Nat. Prod.* 2006, 69, 1065; b) S. Z. Mu, X. W. Yang, Y. T. Di, H. P. He, Y. Wang, Y. H. Wang, L. Li, X. J. Hao, *Chem. Biodiv.* 2007, 4, 129.
- [6] S. Yamamura, J. A. Lamberton, H. Irikawa, Y. Okumura, M. Toda, Y. Hirata, Bull. Chem. Soc. Jpn. 1977, 50, 1836.
- [7] H. Morita, H. Takatsu, J. Kobayashi, Tetrahedron 2003, 59, 3575.
- [8] S. Yamamura, J. A. Lamberton, H. Irikawa, Y. Okumura, Y. Hirata, Chem. Lett. 1975, 9, 923.
- [9] S. Yamamura, M. Toda, Y. Hirata, Bull. Chem. Soc. Jpn. 1976, 49, 839.
- [10] H. E. Bitar, V. H. Nguyen, A. Gramain, T. Sévenet, B. Bodo, J. Nat. Prod. 2004, 67, 1094.
- [11] a) H. Niwa, M. Toda, Y. Hirata, S. Yamamura, *Tetrahedron Lett.* 1972, 13, 2697; b) M. Toda, H. Niwa, H. Irikawa, Y. Hirata, S. Yamamura, *Tetrahedron* 1974, 30, 2683; c) H. Morita, N. Yoshida, J. Kobayashi, *Tetrahedron* 2000, 56, 2641.
- [12] H. Sakurai, H. Irikawa, S. Yamamura, Y. Hirata, Tetrahedron Lett. 1967, 8, 2883.
- [13] C. Q. Fan, S. Yin, J. J. Xue, J. M. Yue, Tetrahedron 2007, 63, 115.
- [14] H. Morita, J. Kobayashi, Org. Lett. 2003, 5, 2895.

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